



# CELL BOUNDARIES

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- Cells create boundaries through:
  - Cell Membranes
  - Cell Walls



# DIFFUSION THROUGH CELL BOUNDARIES

- Every cell = a liquid environment

So why have a cell membrane ?

- Function of the cell membrane = regulate the movement of dissolved molecules from one side of membrane to the other



# DEMO

- Beaker = Liquid outside of a cell
  - Baggie= Cell membrane
  - Inside Baggie = Inside a cell
- 
- Hypothesis ? How will our cell membrane react to the iodine?

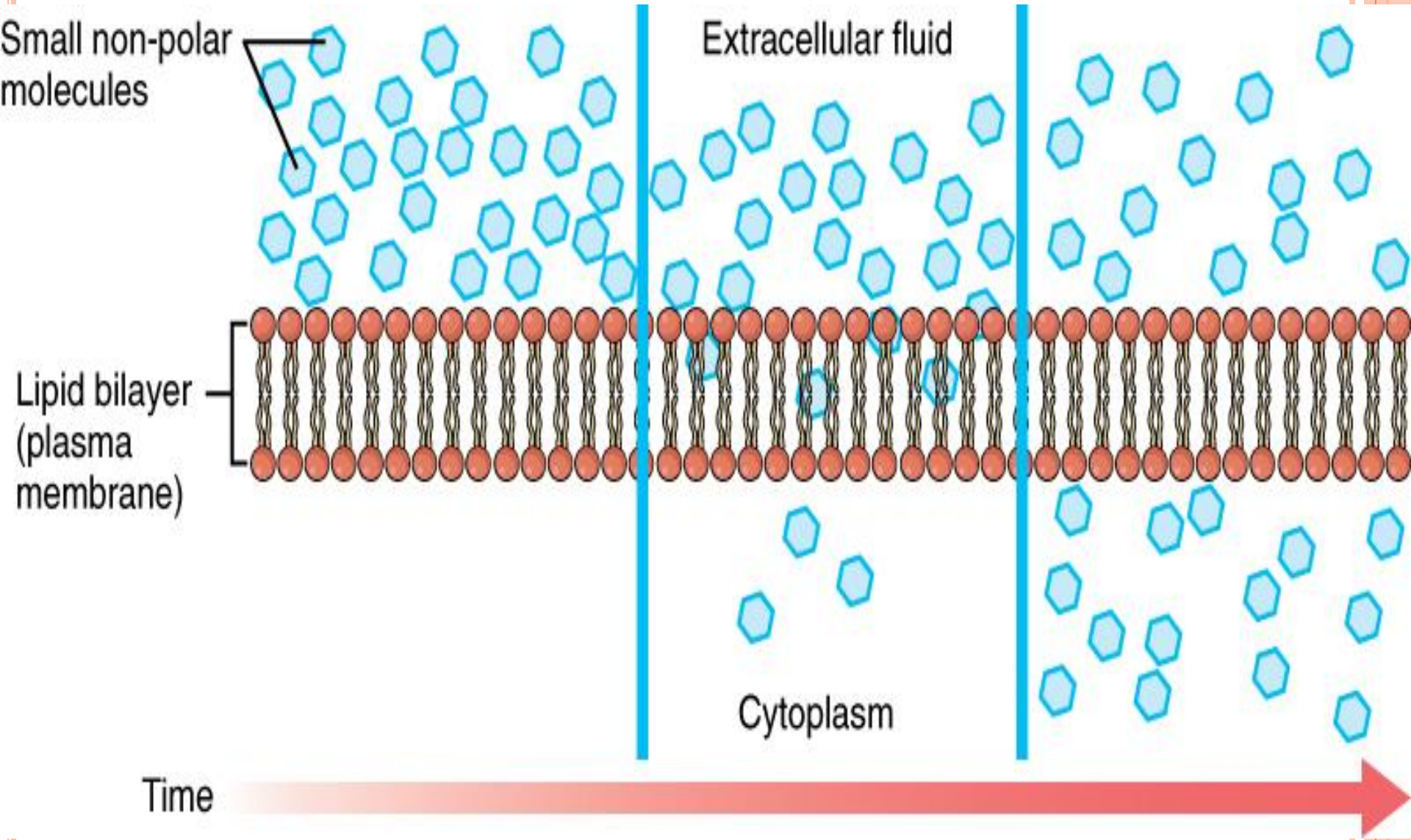


# DIFFUSION

- In a solution, particles = move constantly
- R → particles move from an area of more concentrated to less concentrated
- A process = DIFFUSION.
- When the concentration of the solution = same throughout a system, R → EQUILIBRIUM.



# DIFFUSION



# TYPES OF MEMBRANES

- Some substances = too large or too strongly charged to cross the lipid bilayer.
- Substance can diffuse across a membrane, the membrane = *permeable* to it.
- Substance cannot diffuse = *impermeable* to it.



# OSMOSIS

- Most membranes = selectively permeable

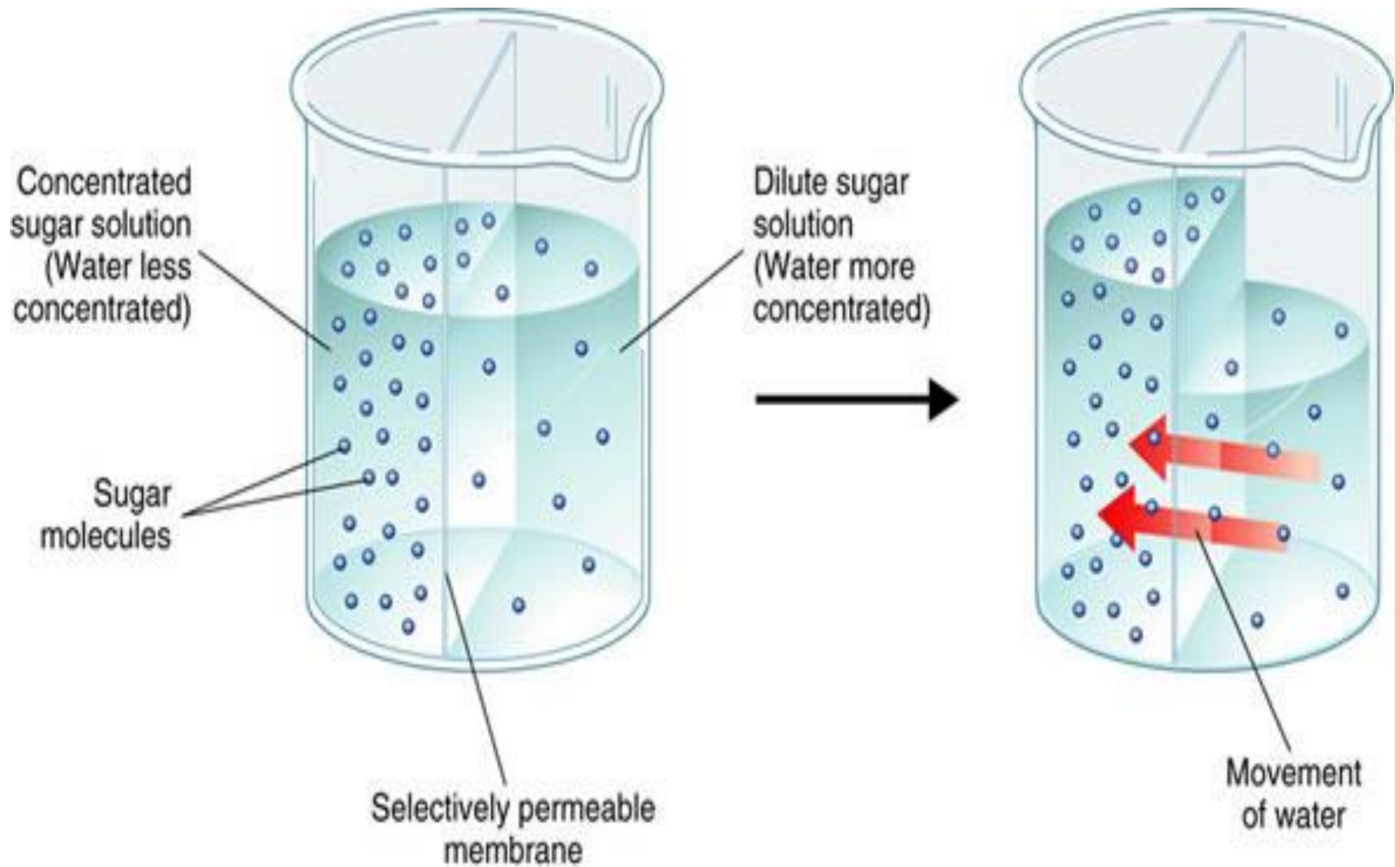
What do you think that means?

Some substances can pass across and others cannot

- Water passes easily, even though many solute molecules cannot.
- R → OSMOSIS
- OSMOSIS: The diffusion of water through a selectively permeable membrane.







# OSMOSIS – DIFFUSION

- They sound similar but they ARE NOT!
- Watch this video
- Make a VENN DIAGRAM that compares the two processes
- Put this in your notes \*\*\* Good Quiz Question  
\*\*\*



What was the point of  
the demonstration  
yesterday?

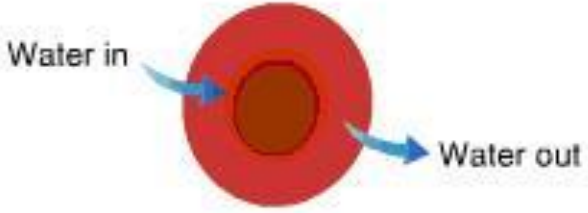



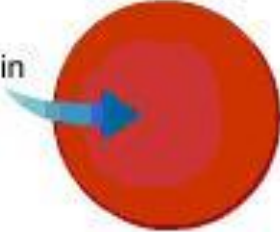



## TYPES OF “FLUID ENVIRONMENTS”

- Isotonic: same concentration inside and outside of cell
- Hypertonic: solution has higher concentration than cell
- Hypotonic: solution has lower concentration than cell



## The Effects of Osmosis on Cells

Solution	Animal Cell	Plant Cell
<p><b>Isotonic:</b> The concentration of solutes is the same inside and outside the cell.</p>		
<p><b>Hypertonic:</b> Solution has a higher solute concentration than the cell.</p>		
<p><b>Hypotonic:</b> Solution has a lower solute concentration than the cell.</p>		

**Effects of Osmosis** Cells placed in an isotonic solution neither gain nor lose water. In a hypertonic solution, animal cells shrink, and plant cell vacuoles collapse. In a hypotonic solution, animal cells swell and burst. The vacuoles of plant cells swell, pushing the cell contents out against the cell wall. Predicting *What would happen to the animal cell in the isotonic solution if it were placed in pure water?*

- Isotonic: Cells neither gain nor lose water.
- Hypertonic: Animal cells shrink while vacuole of plant cells collapse.
- Hypotonic: Animal cells swell and burst while the vacuole of plant cells swell, pushing cell contents out against cell wall.



# OSMOTIC PRESSURE

- Osmosis R-> pressure known as osmotic pressure on the hypertonic side of a selectively permeable membrane.
  
- Ex. Cell and fresh water:
  - I. cell will be hypertonic due to sugars, proteins, etc.
  - II. Osmotic pressure will push water into the cell (trying to equal out the concentration).
  - III. Cell will expand and eventually burst.



- Cells in large organisms are not in danger of bursting.
- Most cells in large organisms don't come in contact with fresh water.
- Cells are bathed in fluids, (blood), that are isotonic ie: same strength as cell





- Other cells, (plant cells & bacteria ), which come into contact w/ fresh water, are surrounded by a tough cell wall.
- Cell walls prevent the cells from expanding and bursting even under osmotic pressure.



# CELL MEMBRANE AND TRANSPORT

- Crash Course

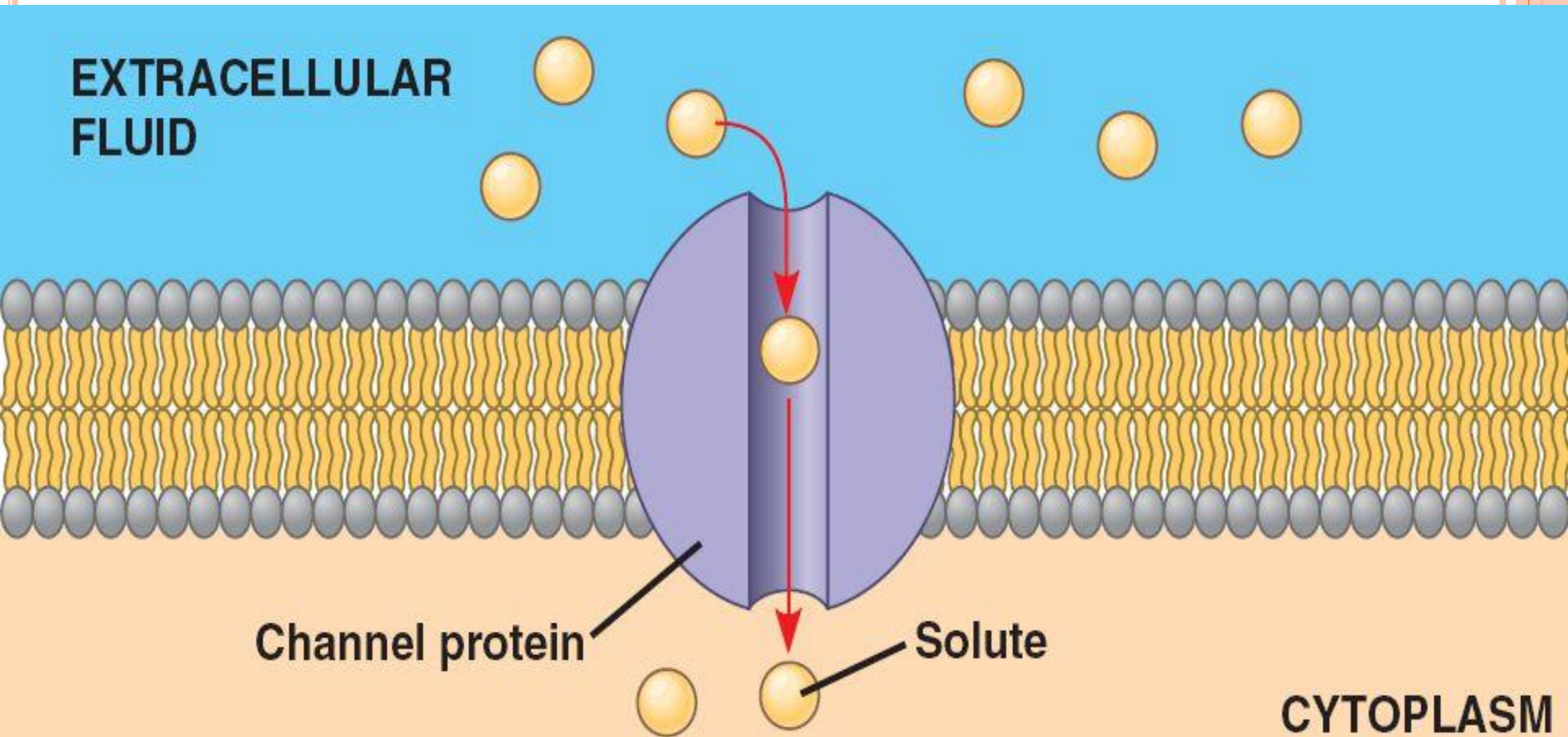


# FACILITATED DIFFUSION

- Water molecules can cross a membrane through the lipid bilayer, but how do other molecules cross?
- Cell membranes have protein channels that act as carriers, making it easy for certain molecules to cross!



**Facilitated diffusion= glucose (cannot diffuse across the lipid bilayer on their own)  
R-> move through protein channels instead.**



# FACILITATED DIFFUSION

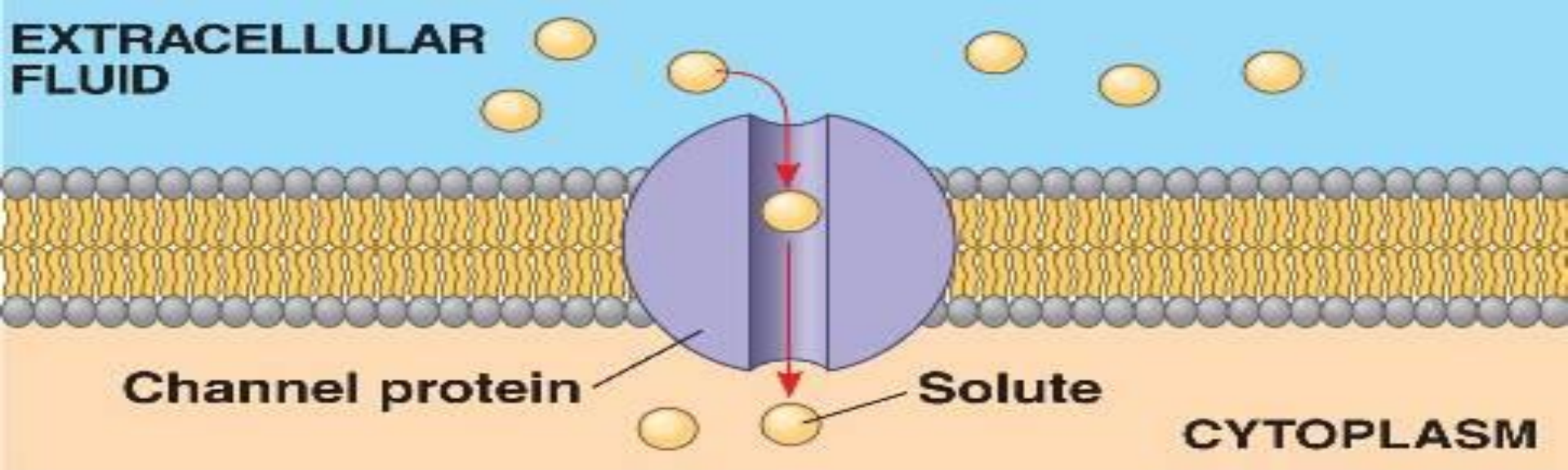
- Hundreds of protein channels allow particular substances to cross different membranes.
- This movement **DOES NOT** require the use of the cell's energy.



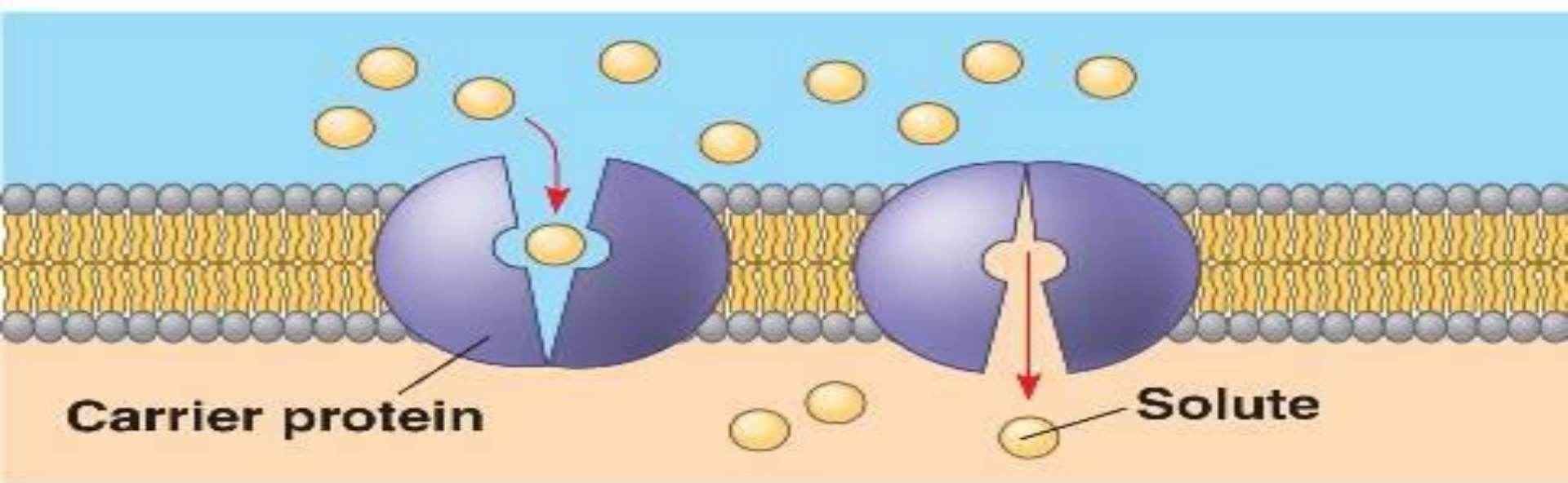
# ACTIVE TRANSPORT

- Cells sometimes must move materials in the opposite direction – against a concentration difference.
- This is done by a process known as **active transport**.
- This requires energy!



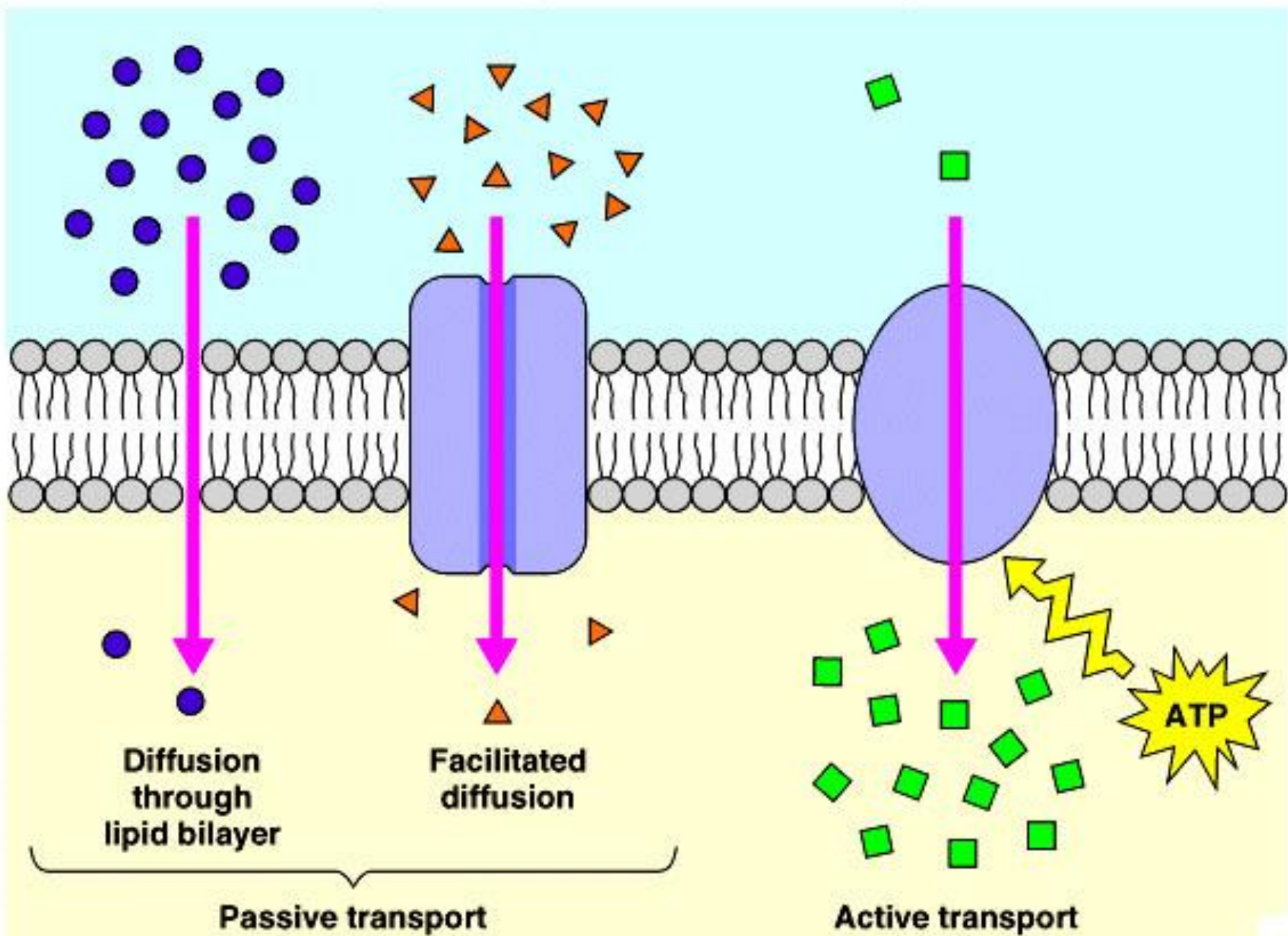


**(a) A channel protein**



**(b) A carrier protein**

Figure 8.14 Review: A comparison of passive and active transport





# ENDOCYTOSIS & EXOCYTOSIS

- The active transport of small molecules / ions across a cell membrane = transport proteins or “pumps” found in the membrane itself.
- Larger molecules and clumps of materials can be actively transported
- endocytosis and exocytosis.



- The transport involves changes in the shape of the cell membrane.



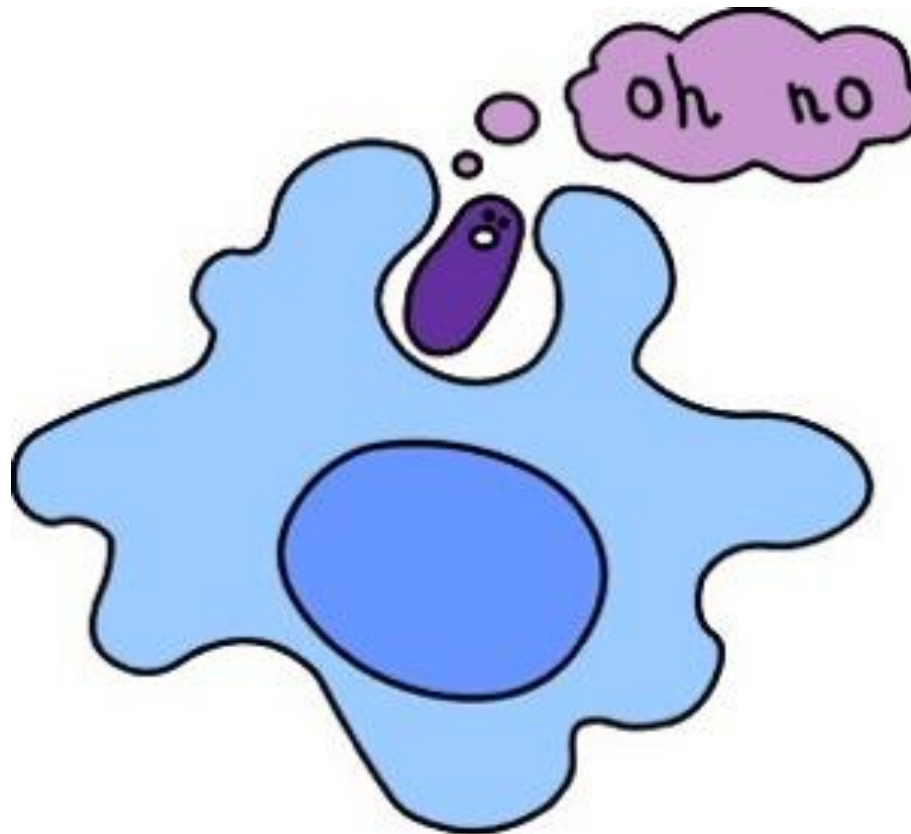
# ENDOCYTOSIS

- Taking materials into the cell by means of pockets of cell membrane.
- The pocket breaks loose from the outer portion of the cell membrane and forms a vacuole in the cytoplasm.



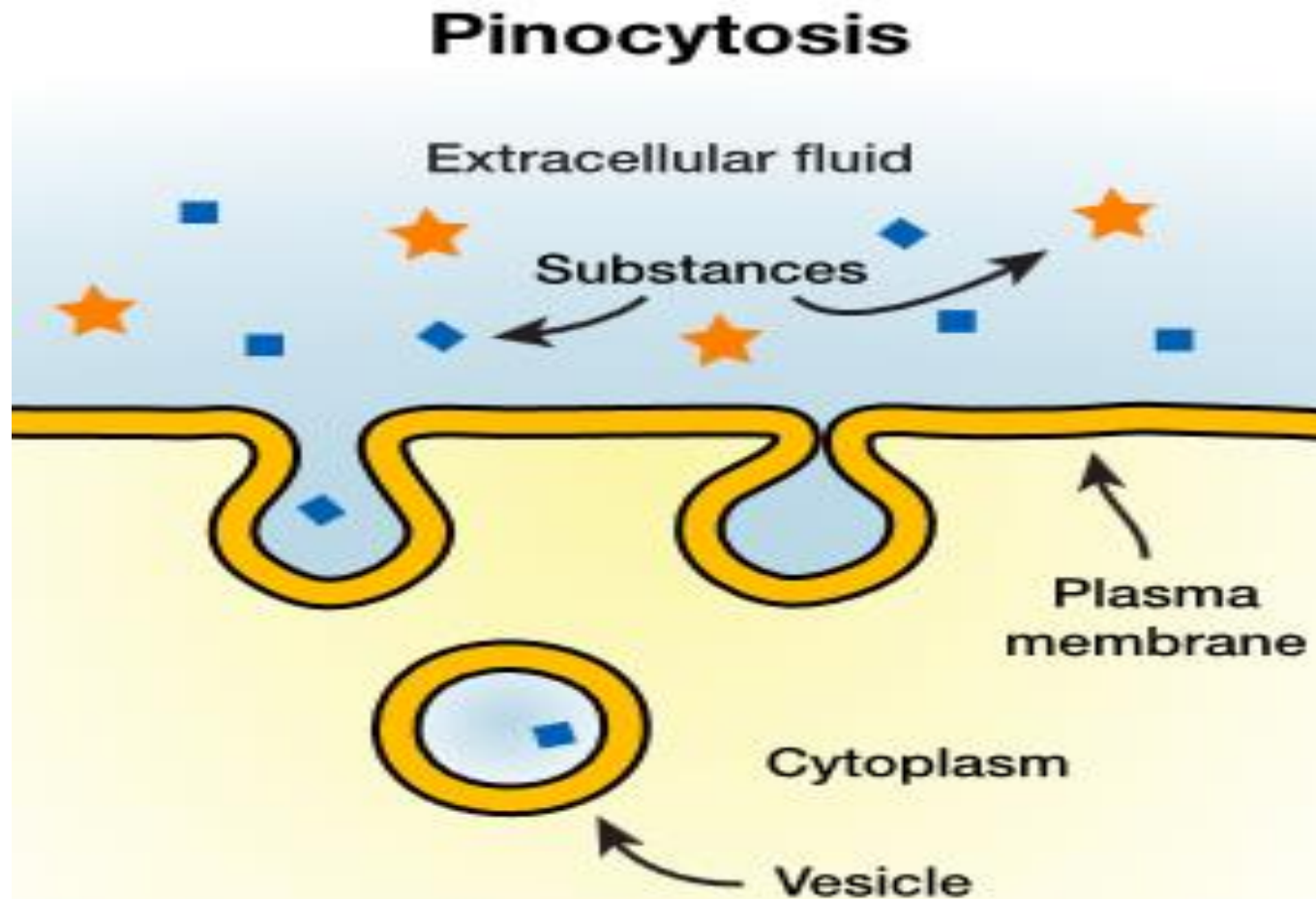
○ One example of Endocytosis is:

1. **Phagocytosis:** cytoplasm surrounds a particle and packages it within a food vacuole. The cell then engulfs it!



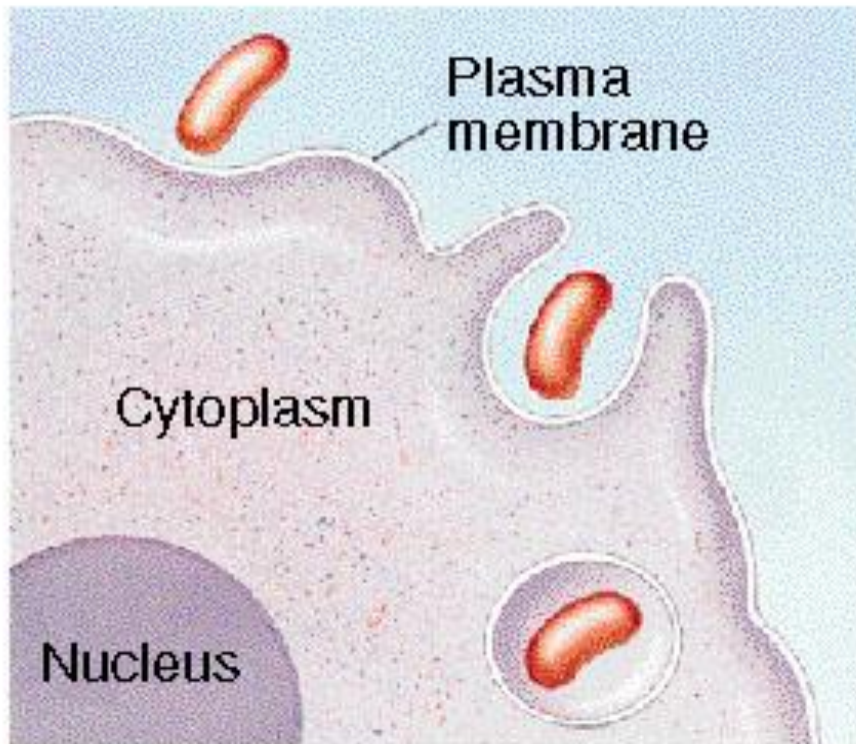
- Another example of Endocytosis is:

**2. Pinocytosis:** Tiny pockets form along cell membrane, fill with fluid and pinch off to form a vacuole within the cell.

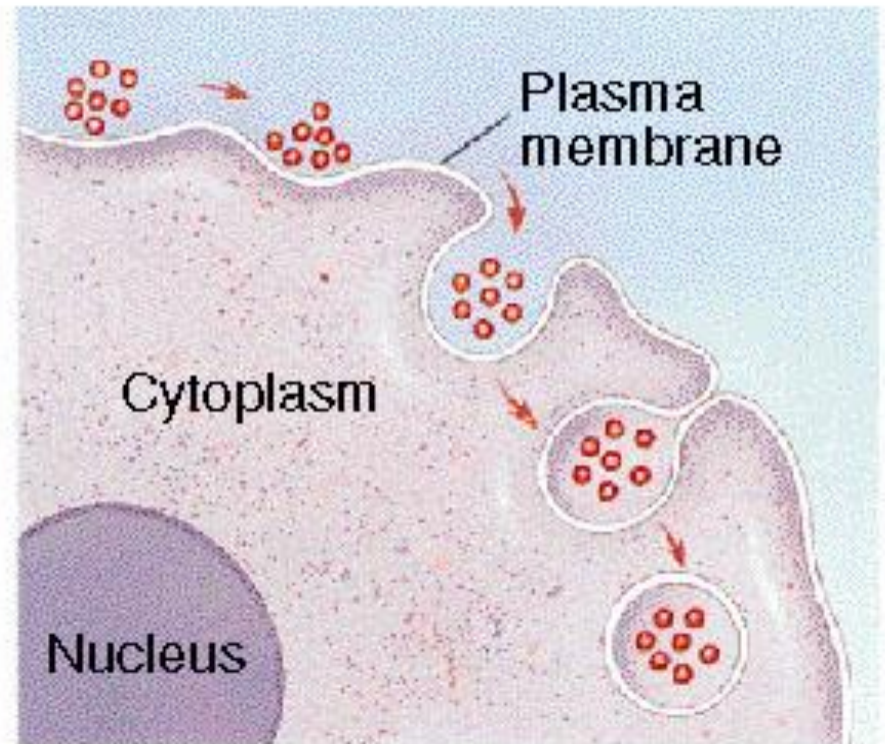


# WHAT'S THE DIFFERENCE?

## Endocytosis



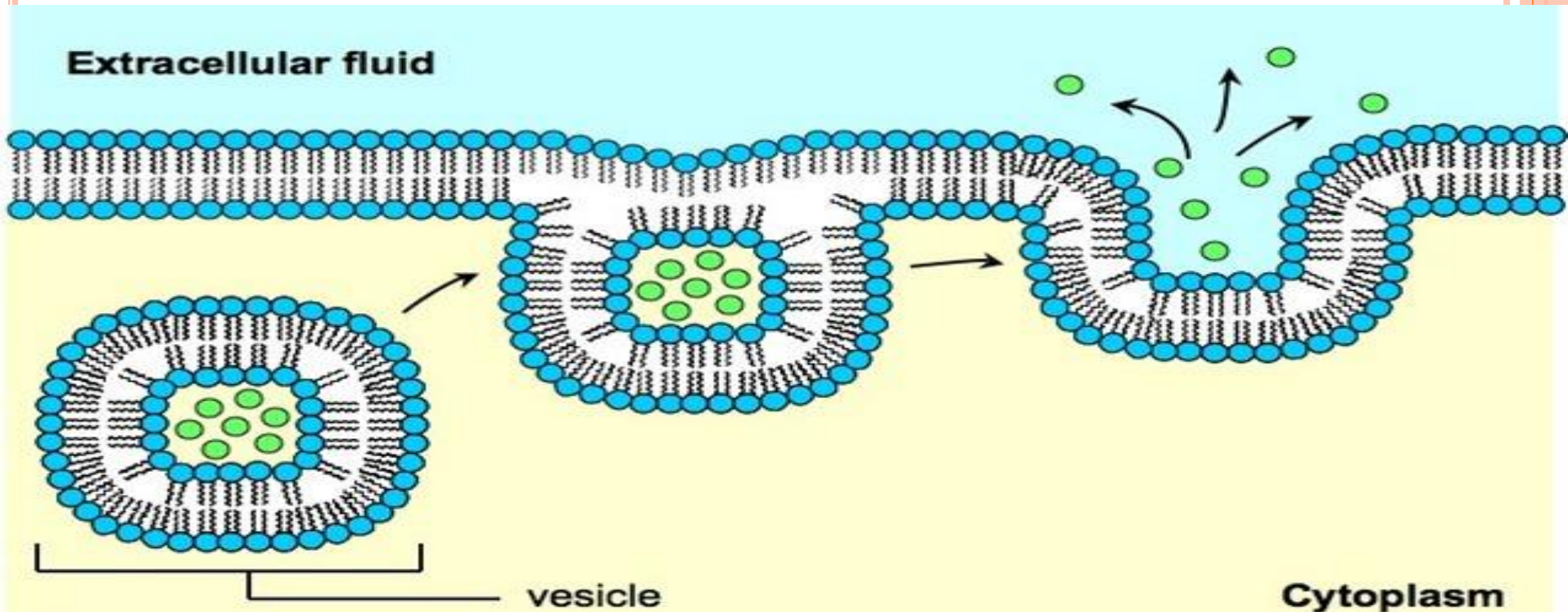
(a) PHAGOCYTOSIS



(d) PINOCYTOSIS

# EXOCYTOSIS

- Large amounts of material released from the cell.
- The membrane of the vacuole surrounding the material fuses with the cell membrane, forcing the contents out of the cell.



# ACTIVITY

- With the items brought into class, I want you to design the three ways substances cross the phospholipid bilayer
- Groups of 2-3
- Each group choose a different manner:
  - Diffusion
  - Facilitated Diffusion
  - Active Transport

